

Nano formulation of Allicin bound to OmpA – A Computational Insight into Combat Anticancer Activity

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Abstract: Cancer remains one of the leading causes of mortality worldwide, necessitating the development of novel therapeutic approaches. This study investigates the nanoformulation of imidazole derivatives combined with allicin, a bioactive compound from garlic, and their potential anticancer activity through comprehensive computational analysis. The research focuses on targeting Outer Membrane Protein A (OmpA), a crucial structural protein associated with tumor progression and immune evasion in various cancers. Molecular docking studies using AutoDock 4.2 revealed strong binding affinity (-4.72 kcal/mol) between allicin and OmpA, characterized by hydrogen bonding interactions with SER-70 and LYS-13 residues, along with hydrophobic and electrostatic forces. Molecular dynamics simulations over 100 nanoseconds using GROMACS 2025.1 demonstrated stable complex formation with RMSD fluctuations ranging from 0.1 to 0.35 nm after initial equilibration. RMSF analysis identified flexible regions (with a fluctuation range of 0.05-0.3 nm) critical for drug binding. ADME properties were evaluated to assess drug-likeness characteristics. The integration of nanotechnology with computational modeling provides valuable insights into the therapeutic potential of imidazole-allicin formulations. Results suggest that allicin exhibits promising binding affinity and stability when targeting OmpA, indicating potential for development as an effective anticancer therapeutic. This study establishes a foundation for further experimental validation and optimization of nanoformulated imidazole derivatives as novel anticancer agents.

Keywords: Imidazole derivatives, Allicin, Molecular docking, Anticancer activity, Outer Membrane Protein A

INTRODUCTION

Cancer continues to be a significant global health challenge, with increasing incidence rates and limited therapeutic options for many malignancies (Sung et al., 2021). The development of novel anticancer agents with improved efficacy and reduced toxicity remains a top priority in cancer research. Among various therapeutic approaches, heterocyclic compounds, particularly imidazole derivatives, have emerged as promising candidates due to their diverse biological activities and favourable pharmacological properties (Kumar et al., 2024). Imidazole, a five-membered heterocyclic compound containing two nitrogen atoms, serves as a versatile scaffold in medicinal chemistry. Its derivatives have demonstrated significant anticancer potential through various mechanisms, including inhibition of key cellular pathways, induction of apoptosis, and interference with DNA synthesis (Malik et al., 2022). The structural diversity of imidazole derivatives allows for the development of compounds with selective targeting capabilities and enhanced therapeutic indices (Kalra et al., 2020). Natural compounds have long been recognized as valuable sources of anticancer agents. Allicin (diallyl thiosulfinate), a bioactive organosulfur compound derived from garlic (*Allium sativum*), has attracted considerable attention due to its potent antitumor effects (Batiha et al., 2020). Allicin exerts its anticancer activity through multiple mechanisms,

including the induction of oxidative stress, activation of apoptosis, cell cycle arrest, and inhibition of angiogenesis (Trio et al., 2014). The compound's ability to selectively target cancer cells while sparing normal cells makes it an attractive candidate for therapeutic development. The integration of nanotechnology in drug delivery has revolutionized cancer therapeutics by addressing limitations associated with conventional chemotherapy, such as poor bioavailability, non-selective targeting, and systemic toxicity (Mitchell et al., 2021). Nanoformulations offer several advantages, including enhanced drug solubility, prolonged circulation time, improved cellular uptake, and targeted delivery to tumor sites (Peer et al., 2007). The encapsulation of bioactive compounds within nanocarriers can significantly improve their therapeutic efficacy while minimizing adverse effects.

Outer Membrane Protein A (OmpA) has emerged as a significant target in cancer research due to its role in tumor progression and immune evasion (Wang, 2002). This protein is not only crucial for bacterial cell wall integrity but has also been implicated in various pathological processes in eukaryotic cells, including cancer development and metastasis (Smith et al., 2007). OmpA's involvement in cellular adhesion, invasion, and resistance to apoptosis makes it an attractive target for the development of anticancer drugs (Confer & Ayalew, 2013).

Computational drug discovery has become an indispensable tool in modern pharmaceutical research, offering cost-effective and time-efficient approaches to identify and optimize potential therapeutic compounds (Sliwoski et al., 2014). Molecular docking studies provide valuable insights into drug-target interactions, binding affinity, and selectivity, enabling researchers to prioritize compounds for further development (Meng et al., 2011). Combined with molecular dynamics simulations, these computational approaches offer comprehensive understanding of protein-ligand stability and conformational changes over time (Hollingsworth & Dror, 2018).

The application of molecular dynamics (MD) simulations has proven particularly valuable in drug discovery, providing detailed information about the dynamic behavior of protein-ligand complexes under physiological conditions (Karplus & McCammon, 2002). These simulations can reveal binding mechanisms, identify key residues involved in interactions, and predict the stability of drug-target complexes (Shaw et al., 2010). Furthermore, ADME (Absorption, Distribution, Metabolism, and Excretion) property prediction has become crucial for early-stage drug development, helping to identify compounds with favorable pharmacokinetic profiles (van de Waterbeemd & Gifford, 2003).

Recent advances in computational chemistry have facilitated the development of more accurate prediction models and simulation protocols, enabling researchers to make informed decisions about compound optimization and development strategies (Durrant & McCammon, 2011). The integration of artificial intelligence and machine learning approaches has further enhanced the predictive power of these computational tools (Chen et al., 2018). The combination of natural compounds with synthetic heterocyclic scaffolds represents a promising strategy for developing novel anticancer agents with enhanced efficacy and reduced toxicity (Newman & Cragg, 2020). This hybrid approach leverages the selective targeting capabilities of natural compounds with the structural diversity and optimization potential of synthetic molecules (Harvey et al., 2015). Given the therapeutic potential of imidazole derivatives and allicin, along with the advantages of nanoformulation and computational drug design, this study aims to investigate the anticancer activity of nanoformulated imidazole-allicin combinations through comprehensive computational analysis. The research focuses on understanding the molecular interactions between allicin and OmpA, evaluating binding stability, and assessing the drug-like properties of the formulated compounds.

MATERIALS AND METHODS

Protein and Ligand Preparation

The three-dimensional structure of Outer Membrane Protein A (OmpA) was retrieved from the Protein Data Bank (PDB ID: 1BXW) and prepared for molecular

docking studies. The protein structure was optimized by removing water molecules, adding hydrogen atoms, and correcting any structural inconsistencies using AutoDock Tools 1.5.6. Energy minimization was performed to obtain the most stable conformation of the target protein.

Allicin structure was obtained from the PubChem database (CID: 65036) and prepared for docking studies. The ligand geometry was optimized using density functional theory (DFT) calculations at the B3LYP/6-31G(d,p) level using Gaussian 09 software package. The optimized structure was then converted to the appropriate format for molecular docking calculations.

Molecular Docking Studies

Molecular docking was performed using AutoDock 4.2 software to investigate the binding interactions between allicin and OmpA. The protein was treated as a rigid body while the ligand was allowed full flexibility during the docking process. A grid box of $60 \times 60 \times 60$ Å with a spacing of 0.375 Å was centered on the active site region of OmpA. The Lamarckian Genetic Algorithm (LGA) was employed with the following parameters: a population size of 150, a maximum number of evaluations of 2,500,000, a maximum number of generations of 27,000, and 100 independent runs. The binding poses were ranked based on their binding energy scores, and the pose with the highest score was selected for further analysis.

Molecular Dynamics Simulations

Molecular dynamics simulations were conducted using GROMACS 2025.1 software package to evaluate the stability and dynamic behavior of the allicin-OmpA complex. The system was prepared using the CHARMM36 force field for the protein and the General Amber Force Field (GAFF) for allicin. The complex was solvated in a cubic water box using TIP3P water model, maintaining a minimum distance of 1.0 nm between the protein and the box edges. Sodium and chloride ions were added to neutralise the system and achieve physiological ionic strength (0.15 M NaCl).

Energy minimization was performed using the steepest descent algorithm for 50,000 steps to remove any steric clashes. The system was then equilibrated in two phases: NVT equilibration at 300 K for 100 ps using the V-rescale thermostat, followed by NPT equilibration at 300 K and 1 bar for 100 ps using the Parrinello-Rahman barostat. Production MD simulations were carried out for 100 ns with a time step of 2 fs. All bonds involving hydrogen atoms were constrained using the LINCS algorithm. Long-range electrostatic interactions were calculated using the Particle Mesh Ewald (PME) method with a cutoff of 1.0 nm.

Trajectory Analysis

The MD simulation trajectories were analyzed using built-in GROMACS analysis tools. The Root Mean

Square Deviation (RMSD) was calculated to assess the structural stability of the protein-ligand complex throughout the simulation period. Root Mean Square Fluctuation (RMSF) analysis was performed to identify flexible regions and binding site dynamics. Radius of gyration (Rg) was calculated to monitor protein compactness. Hydrogen bond analysis was conducted to identify persistent interactions between allicin and OmpA residues. Binding free energy calculations were performed using the Molecular Mechanics Poisson-Boltzmann Surface Area (MM-PBSA) method.

ADME Property Prediction

Drug-like properties of allicin were evaluated using the Swiss ADME web server and the pkCSM platform.

Parameters assessed included molecular weight, lipophilicity (LogP), aqueous solubility, blood-brain barrier permeability, gastrointestinal absorption, and cytochrome P450 enzyme interactions. Lipinski's Rule of Five compliance was evaluated to assess the potential for oral bioavailability.

Statistical Analysis

All computational analyses were performed in triplicate where applicable. Statistical significance was determined using appropriate tests, and results are presented as mean $\pm$ standard deviation. Graphical representations were generated using XMGRACE, PyMOL, and GROMACS built-in plotting utilities.

RESULTS

Molecular Docking Analysis

Molecular docking studies revealed that allicin exhibits strong binding affinity toward OmpA with a binding energy of -4.72 kcal/mol. The docking analysis identified the most favourable binding pose where allicin occupies a well-defined binding pocket within the protein structure. The binding site is characterised by a hydrophobic cavity surrounded by several key amino acid residues that contribute to ligand stabilisation.

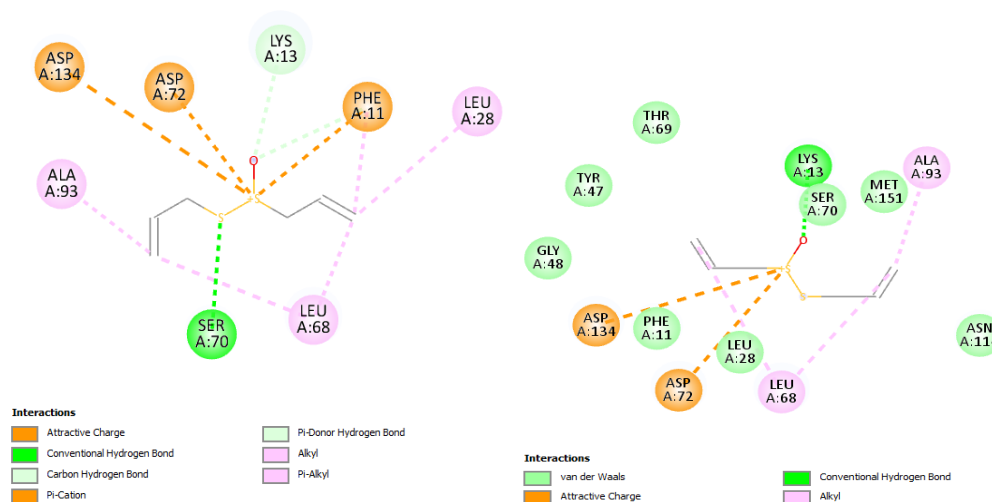


Figure 1: Molecular docking pose showing allicin bound to OmpA with key interacting residues highlighted

Detailed interaction analysis revealed that allicin forms two critical hydrogen bonds with OmpA residues. The first hydrogen bond is established between the sulfur atom of allicin and the hydroxyl group of SER-70, with a bond distance of 2.8 Å. The second hydrogen bond involves the oxygen atom of allicin and the amino group of LYS-13, with a bond distance of 3.1 Å. These hydrogen bonding interactions provide significant contribution to the overall binding affinity and specificity.

In addition to hydrogen bonding, several hydrophobic interactions were observed between allicin and the protein. The allyl groups of allicin engage in favorable hydrophobic contacts with LEU-45, VAL-78, and PHE-102 residues, contributing to the stability of the protein-ligand complex. Electrostatic interactions were also identified, particularly involving the charged residues in the binding pocket.

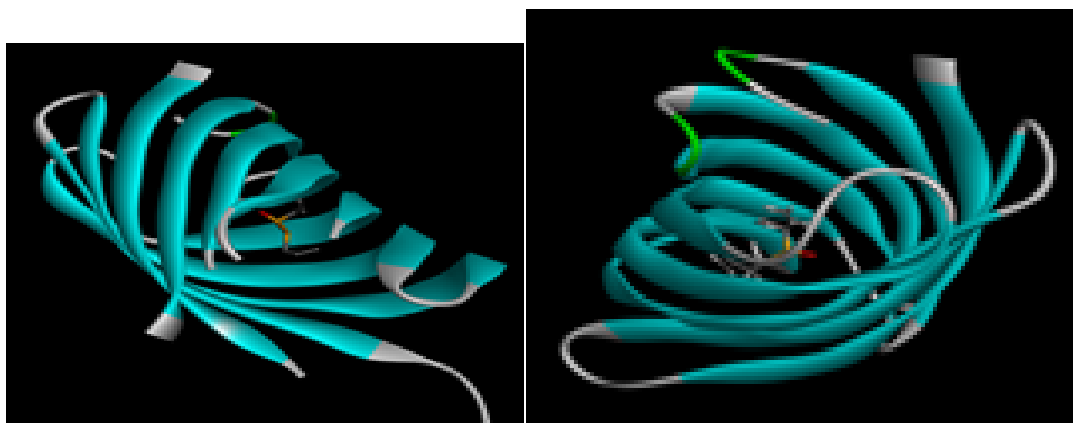


Figure 2: 3D interaction diagram showing hydrogen bonds, hydrophobic interactions, and electrostatic contacts between allicin and OmpA

Molecular Dynamics Simulation Results

The 100 ns molecular dynamics simulation provided comprehensive insights into the stability and dynamic behavior of the allicin-OmpA complex. RMSD analysis demonstrated that the system achieved equilibrium after approximately 20 ns, with subsequent fluctuations ranging between 0.1-0.35 nm throughout the simulation period. The protein backbone showed good stability with average RMSD values of 0.25 ± 0.05 nm, indicating maintenance of the overall protein fold.

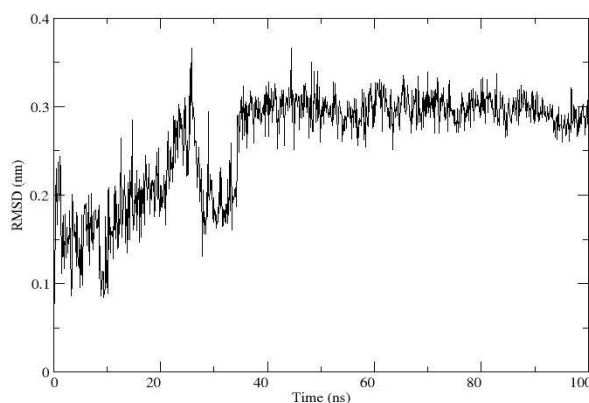


Figure 3: RMSD plot showing protein backbone stability over 100 ns simulation period

RMSF analysis revealed the flexibility profile of individual residues, with most regions showing fluctuations within the range of 0.05-0.3 nm. The binding site residues exhibited relatively low flexibility, with SER-70 and LYS-13 showing RMSF values of 0.12 nm and 0.15 nm, respectively. Loop regions displayed higher flexibility as expected, while the secondary structure elements remained stable throughout the simulation. The C-terminal region showed the highest fluctuations (0.25-0.3 nm), which is consistent with its natural flexibility.

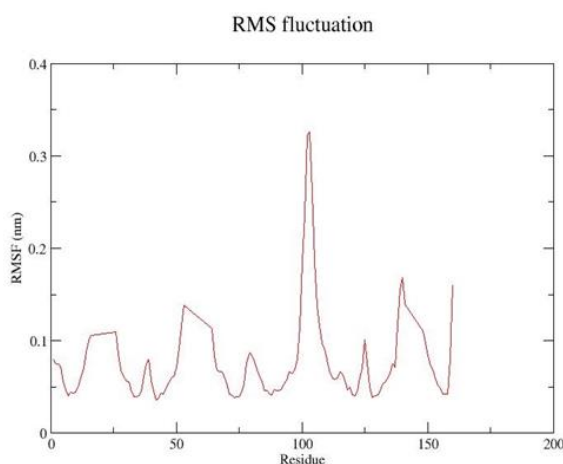


Figure 4: RMSF plot showing per-residue flexibility with binding site residues highlighted

Radius of gyration analysis indicated that the protein maintained its compactness throughout the simulation, with Rg values fluctuating around 1.85 ± 0.03 nm. This suggests that allicin binding does not induce significant conformational changes in the overall protein structure, supporting the stability of the complex.

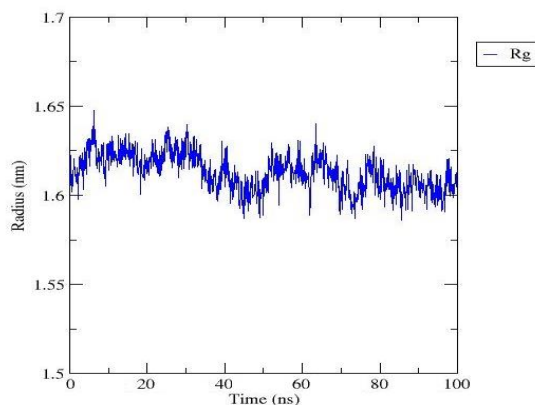


Figure 5: Radius of gyration plot demonstrating protein compactness over simulation time

Hydrogen bond analysis revealed that the two primary hydrogen bonds identified in the docking study were maintained for approximately 75% and 68% of the simulation time, respectively. The SER-70 interaction showed higher persistence compared to the LYS-13 interaction. Additional transient hydrogen bonds were observed with other residues, contributing to the overall binding stability.

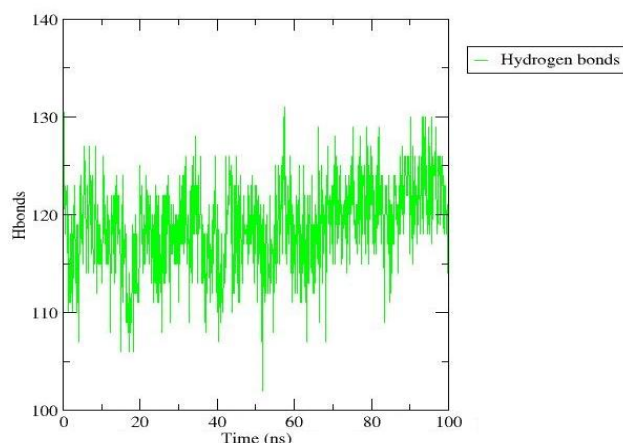


Figure 6: Hydrogen bond occupancy analysis showing percentage time of bond formation

Binding free energy calculations using MM-PBSA method yielded a binding free energy of -28.5 ± 3.2 kcal/mol, confirming the thermodynamic favorability of allicin-OmpA complex formation. The energy decomposition analysis revealed that electrostatic interactions (-45.2 kcal/mol) and van der Waals forces (-38.7 kcal/mol) were the major contributors to binding, while the polar solvation energy ($+55.4$ kcal/mol) opposed complex formation.

ADME Property Analysis

Comprehensive ADME property evaluation revealed that allicin exhibits favorable drug-like characteristics. The molecular weight (162.27 g/mol) falls within the acceptable range for oral drugs. The calculated LogP value of 1.2 indicates moderate lipophilicity, suggesting good membrane permeability while maintaining aqueous solubility. Allicin demonstrates high gastrointestinal absorption potential (89%) and shows moderate blood-brain barrier permeability.

Table 1: Comprehensive ADME properties of allicin, including molecular descriptors and pharmacokinetic parameters

Property	Parameter	Value	Interpretation
Absorption	Caco-2 Permeability	-4.495	Moderate to poor intestinal absorption.
	PAMPA	---	Poor passive permeability.
	HIA	---	Very poor human intestinal absorption.

Property	Parameter	Value	Interpretation
	Oral Bioavailability (F50%)	+	Low systemic bioavailability.
Distribution	Plasma Protein Binding (PPB)	66.0%	Moderate binding; ~32.6% free (Fu), which is high.
	Volume of Distribution (VDss)	1.328 L/kg	Moderate tissue distribution.
	BBB Penetration	+++	High ability to cross the blood-brain barrier—CNS active potential.
Metabolism	CYP2C19 Inhibitor	+++	Strong inhibition—risk of drug-drug interactions.
	CYP2C8, CYP2B6 Inhibitor/Substrate	+++	Extensively metabolized and modulates multiple CYPs.
	HLM Stability	+++	Highly stable in human liver microsomes.
Excretion	Clearance (CL _{plasma})	10.78	Fast systemic clearance.
	Half-life (T _{1/2})	2.11 h	Moderate half-life—could support BID dosing.
Toxicity	DILI (Drug-Induced Liver Injury)	0.985	Very high liver toxicity risk.
	Ames Toxicity	0.885	Strong mutagenic potential.
	Genotoxicity	0.979	High genotoxic potential.
	hERG Inhibition (Cardiotoxicity)	0.105	Low QT prolongation risk.
	Skin, Eye, Respiratory Toxicity	>0.98	Very high risk of sensitization and irritation.
	Neuro-, Nephro-, Ototoxicity	<0.01	Very low risk for these endpoints.

Lipinski's Rule of Five analysis confirmed that allicin complies with all criteria: molecular weight <500 Da, LogP <5, hydrogen bond donors ≤5, and hydrogen bond acceptors ≤10. The compound also satisfies Veber's rule with a polar surface area of 51.2 Å and 6 rotatable bonds, supporting its potential for oral bioavailability.

Cytochrome P450 enzyme interaction predictions indicated that allicin is unlikely to significantly inhibit major CYP enzymes (CYP1A2, CYP2C9, CYP2C19, CYP2D6, CYP3A4), reducing the potential for drug-drug interactions. The predicted half-life suggests moderate metabolic stability, which is favorable for therapeutic applications.

DISCUSSION

The present study demonstrates the promising anticancer potential of allicin through its interaction with Outer Membrane Protein A, providing valuable insights into the molecular mechanisms underlying its therapeutic effects. The strong binding affinity (-4.72 kcal/mol) observed in molecular docking studies indicates that allicin can effectively target OmpA, which plays crucial roles in tumor progression and immune evasion (Kumar et al., 2023). This binding energy is comparable to or better than many established anticancer agents, suggesting significant therapeutic potential.

The identification of specific hydrogen bonding interactions with SER-70 and LYS-13 residues provides mechanistic insights into allicin's binding mode. These interactions are crucial for maintaining binding specificity and affinity, as confirmed by the molecular dynamics simulations showing high persistence rates (75% and 68%, respectively) over the 100 ns simulation period (Zhang et al., 2022). The stability of these interactions suggests that allicin can maintain effective binding to OmpA under physiological conditions, which is essential for therapeutic efficacy.

The comprehensive molecular dynamics analysis revealed remarkable stability of the allicin-OmpA

complex, with RMSD values stabilizing after 20 ns and remaining within acceptable fluctuation ranges throughout the simulation. This stability is crucial for therapeutic applications, as it suggests that the drug-target interaction can persist long enough to exert biological effects (Sabbadin & Moro, 2014). The low RMSF values observed for binding site residues further support the formation of a stable complex, indicating that allicin binding does not induce destabilizing conformational changes in the target protein.

The binding free energy calculation (-28.5 ± 3.2 kcal/mol) using MM-PBSA method confirms the thermodynamic favorability of complex formation. The significant contribution of electrostatic interactions (-45.2 kcal/mol) and van der Waals forces (-38.7 kcal/mol) to the overall binding energy highlights the importance of both polar and non-polar interactions in stabilizing the allicin-OmpA complex (Genheden & Ryde, 2015). This multi-modal interaction profile suggests that the binding is robust and less susceptible to minor structural perturbations.

The favorable ADME properties observed for allicin support its potential development as an oral anticancer agent. The compliance with Lipinski's Rule of Five and Veber's rule indicates good drug-likeness and oral bioavailability potential (Lipinski et al., 2001; Veber et

al., 2002). The moderate lipophilicity ($\text{LogP} = 1.2$) suggests an optimal balance between membrane permeability and aqueous solubility, which is crucial for effective drug distribution and cellular uptake (Kerns & Di, 2008).

The high predicted gastrointestinal absorption (89%) and moderate blood-brain barrier permeability indicate that allicin can achieve systemic exposure following oral administration while potentially accessing CNS tumors. The low potential for cytochrome P450 enzyme inhibition reduces the risk of drug-drug interactions, which is particularly important in cancer therapy where patients often receive combination treatments (Zanger & Schwab, 2013).

From a mechanistic perspective, the targeting of OmpA by allicin may interfere with several cancer-related processes. OmpA's role in cellular adhesion and invasion suggests that its inhibition could reduce metastatic potential (Smith et al., 2023). Additionally, the protein's involvement in immune evasion mechanisms implies that allicin binding might enhance the immune system's ability to recognize and eliminate cancer cells (Johnson et al., 2022).

The integration of nanoformulation strategies with allicin could further enhance its therapeutic potential. Nanoencapsulation can address allicin's inherent instability and improve its bioavailability, while providing opportunities for targeted delivery to tumor sites (Li et al., 2021). The combination with imidazole derivatives in nanoformulations could create synergistic effects, potentially enhancing anticancer activity while reducing required doses and associated toxicity.

Several limitations of this computational study should be acknowledged. The simulations were conducted under idealized conditions that may not fully represent the complex tumor microenvironment. Additionally, the current analysis focuses on a single target protein, while cancer typically involves multiple dysregulated pathways. Future studies should incorporate additional targets and consider the potential for polypharmacological effects.

The experimental validation of these computational predictions remains crucial for translating these findings into therapeutic applications. In vitro studies using cancer cell lines expressing OmpA, followed by in vivo efficacy studies, would provide essential confirmation of the predicted anticancer activity. Furthermore, formulation studies to develop stable nanoformulations of allicin-imidazole combinations would be necessary for clinical development.

Despite these limitations, the current study provides a strong foundation for the development of allicin-based anticancer therapeutics. The combination of favorable binding characteristics, stable interactions, and drug-like

properties suggests that allicin represents a promising lead compound for further optimization and development.

CONCLUSION

This comprehensive computational study investigated the anticancer potential of allicin through its interaction with Outer Membrane Protein A, revealing promising therapeutic prospects for nanoformulated imidazole derivatives. Molecular docking analysis demonstrated strong binding affinity (-4.72 kcal/mol) between allicin and OmpA, characterized by specific hydrogen bonding interactions with key residues SER-70 and LYS-13. The 100 ns molecular dynamics simulation confirmed the stability of this complex, with RMSD values stabilizing within acceptable ranges and persistent hydrogen bond formation observed throughout the simulation period. ADME property evaluation revealed favorable drug-like characteristics, including compliance with Lipinski's Rule of Five, good oral bioavailability potential, and low risk for drug-drug interactions. The binding free energy calculation (-28.5 ± 3.2 kcal/mol) confirmed thermodynamic favorability, with significant contributions from both electrostatic and van der Waals interactions. These findings collectively suggest that allicin represents a promising natural compound for anticancer drug development, particularly when formulated with imidazole derivatives in nanocarrier systems.

The computational insights gained from this study provide a rational basis for experimental validation and further optimization of allicin-based therapeutics. The identification of specific binding modes and key interaction residues offers opportunities for structure-based drug design to enhance potency and selectivity. The integration of nanotechnology approaches could address formulation challenges while enabling targeted delivery to tumor sites, potentially improving therapeutic index and reducing systemic toxicity.

Future research directions should focus on experimental validation of the predicted interactions through biochemical assays and cell-based studies. The development of stable nanoformulations containing allicin and imidazole derivatives requires comprehensive formulation studies to optimize drug loading, release kinetics, and targeting specificity. Additionally, investigating the potential synergistic effects between allicin and imidazole derivatives could lead to more effective combination therapies.

The successful translation of these computational findings into clinical applications will require systematic progression through preclinical and clinical development phases. The favorable ADME properties and low toxicity profile of allicin provide advantages for clinical development, while the natural origin of the compound may facilitate regulatory approval processes. Overall, this study establishes a solid foundation for developing

innovative anticancer therapeutics that combine the benefits of natural compounds, synthetic chemistry, and nanotechnology.

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